Non-equilibrium calcium ionisation in the solar atmosphere

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in collaboration with M. Carlsson

INTRODUCTION CALCIUM

- <u>Chemical abundance:</u>
 - more than 5 orders of magnitude less abundant than hydrogen
- <u>lonisation stage</u> in the (quiet) solar atmosphere:
 - mostly singly ionised (Ca II) [partially ionised]
- Calcium is a "<u>minority species</u>" and only minor electron donor

BUT:

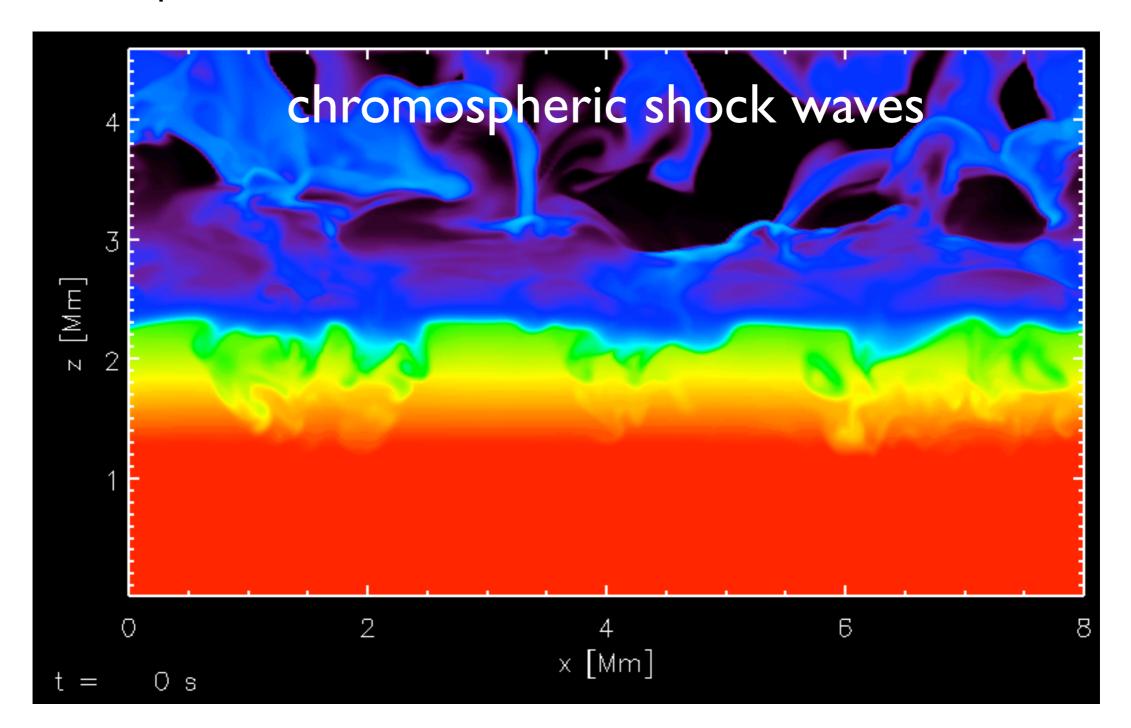
- Significant part of the <u>radiative losses</u> in the chromosphere due to Ca lines
- Spectral lines of Ca II important and widely used <u>diagnostics</u> for chromospheric plasma
- Understanding properties of Ca needed for a detailed quantitative interpretation of observation and derivation of atmospheric properties

INTRODUCTION CALCIUM

- Detailed numerical non-equilibrium treatment of a model atom (incl. time-dependence) can be computationally expensive.
- Modelling of calcium in the solar chromosphere mostly simplified:
 - I-D static (VAL-type)
 - instantaneous equilibrium in 2-D/3-D simulations (often only indirectly included in equation of state, opacities)

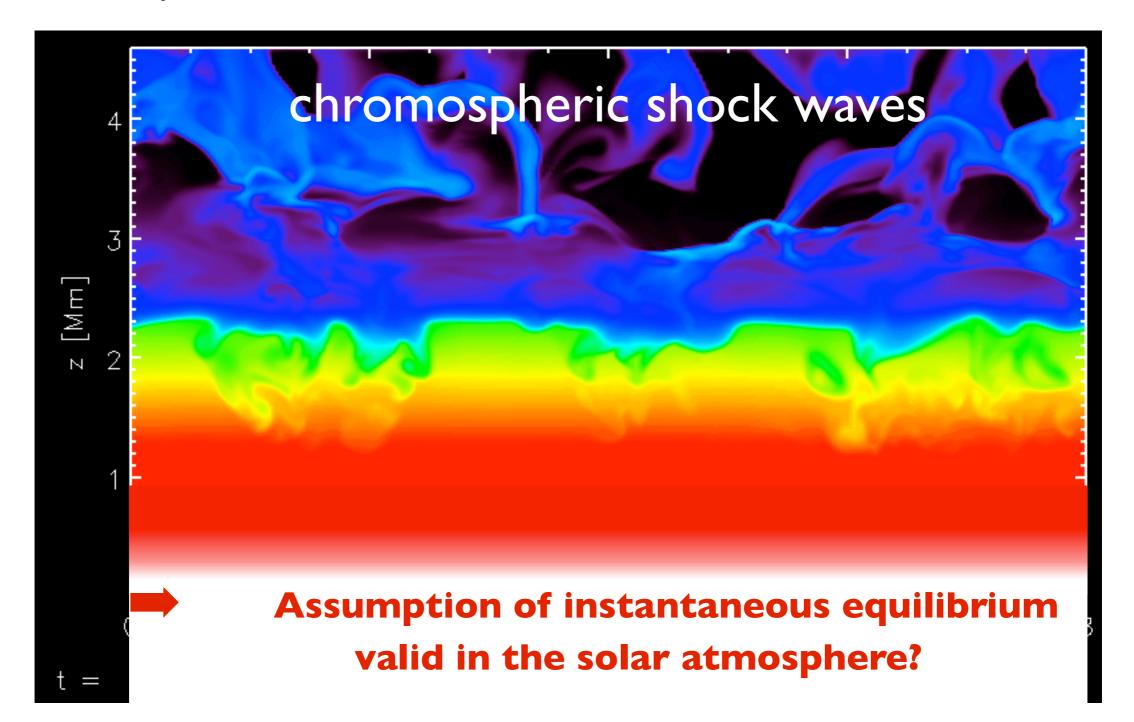
INTRODUCTION

• **BUT:** Solar chromosphere varies strongly on small spatial and temporal scales.



INTRODUCTION

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Wedemeyer-Böhm & Carlsson 2011, A&A 528, A1

- Numerical simulations of the solar chromosphere with detailed time-dependent treatment of Ca ionisation
 - computationally expensive
 - ► I-D first, no magnetic fields(!)
 - similar to a study of the H ionisation (Carlsson & Stein 2002, ApJ, 572, 626)
- Presented here:
 - Resulting effect on the ionisation fraction of Ca
 - Relevant time scales

Simulation code RADYN (Carlsson & Stein 1995)

- solves
 - equations of mass, momentum, energy, and charge conservation
 - non-LTE radiative transfer
 - population rate equations
- takes into account non-equilibrium ionisation, excitation, and radiative energy exchange from the atomic species <u>H, He, and Ca</u> with back-coupling on the hydrodynamics
- adaptive mesh in one spatial dimension
- in the output:
 - population densities of all considered atomic levels

Simulation code RADYN (Carlsson & Stein 1995)

• lower and upper boundaries are both transmitting.

lower boundary

- located in the convection zone (z=-480 km below $\tau 500 = 1$)
- piston excites waves
- piston velocity based on observed Doppler-shifts (Fel line at $\lambda = 396.68$ nm, Lites et al. 1993)

upper boundary condition

- at a height of 10 Mm
- represents a corona on top of the simulated layers
 (T set to 10⁶ K + incident radiation; Tobiska 1991)

Simulation code RADYN (Carlsson & Stein 1995)

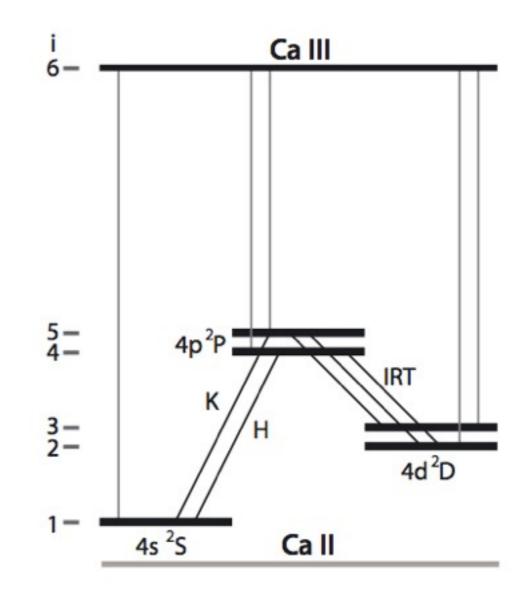
Model atoms:

- 6-level atoms for H and Ca
- 9-level atom for He (doubly ionised He included)
- Transitions between all considered atomic levels treated in detail:
 - Radiative bound-bound transitions: 31–101 frequency points
 - Radiative bound-free transitions: 4–23 frequency points
- All other elements enter as background continua in LTE (derived with the Uppsala atmospheres program; Gustafsson 1973)

NUMERICAL SIMULATIONS THE CALCIUM MODEL ATOM

Energy levels:

- 5 bound states of singly ionised Ca (Ca II)
 - lowest energy level (i = 1) is the ground state of Ca II (4s ²S)
 - two important energy level pairs
 - $4p^{2}P$ (*i* = 2, 3)
 - 4d ²D (*i* = 4, 5)
- continuum level (i = 6)
 - next ionisation stage (Ca III).

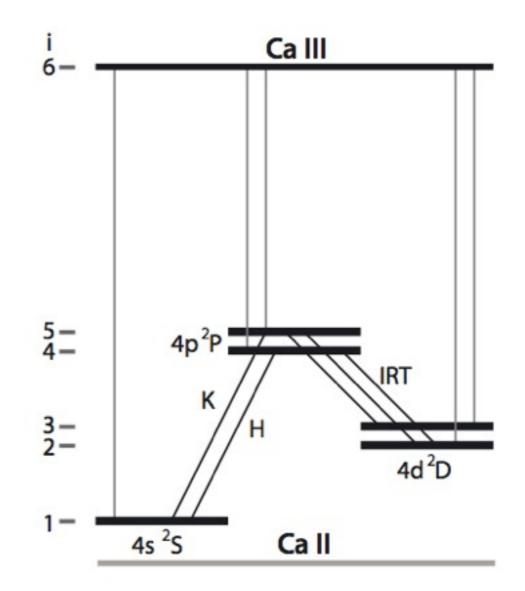


NON-EQUILIBRIUM CALCIUM IONISATION

NUMERICAL SIMULATIONS THE CALCIUM MODEL ATOM

Radiative transitions:

- all 5 allowed radiative bound-bound (b-b) transitions:
 - H + K resonance lines
 - infrared triplet (IRT)
- all 5 radiative bound-free (b-f) transitions
 with corresponding
 photoionisation continua
 from the 5 lowest levels to
 continuum level *i* = 6



- (TD) Detailed time-dependent simulation
- (SE) large number of additional short simulation runs, starting from a statistical equilibrium solution (used for timescale analysis)

IONISATION FRACTION

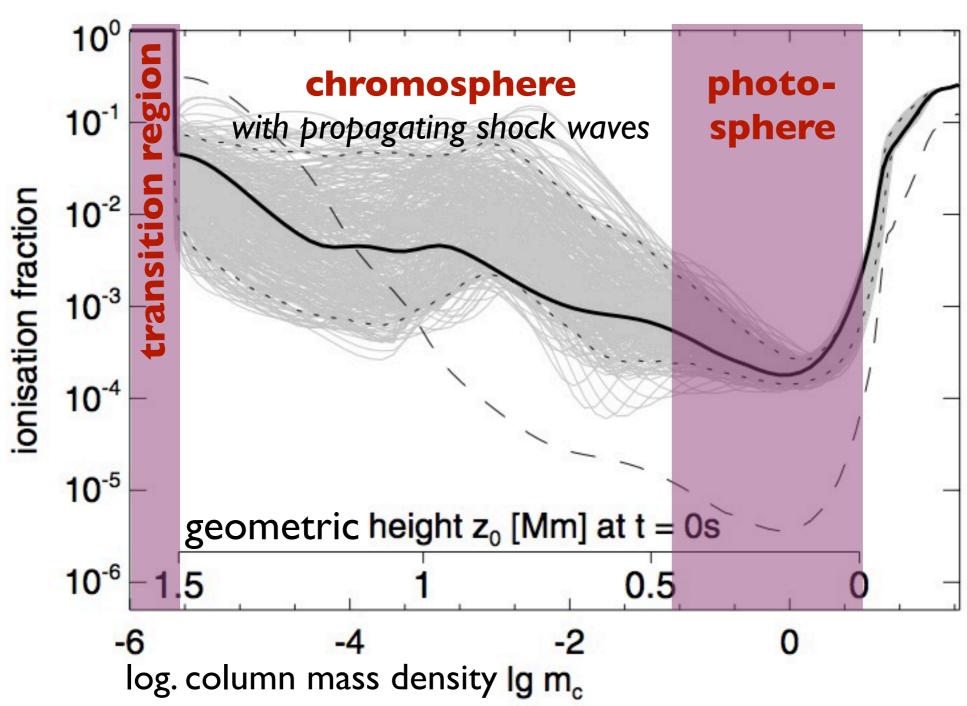
• ionisation fraction

$$\chi_{Ca} = \frac{n_{CaIII}}{n_{CaII} + n_{CaIII}} = \frac{n_6}{\sum_{i=1}^6 n_i}$$

n_i : population densities of the energy levels

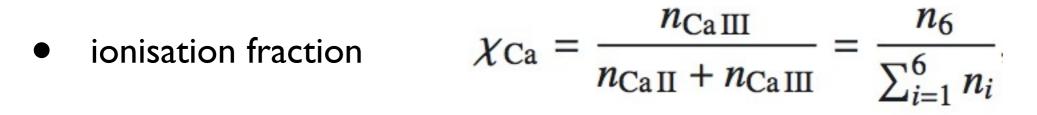
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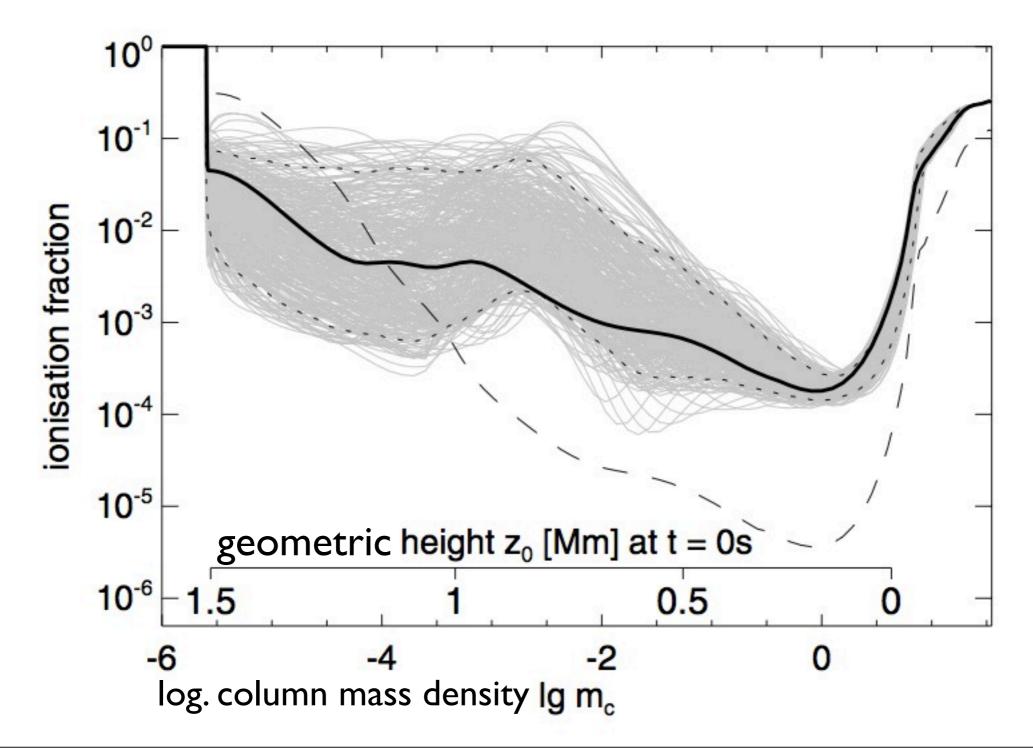
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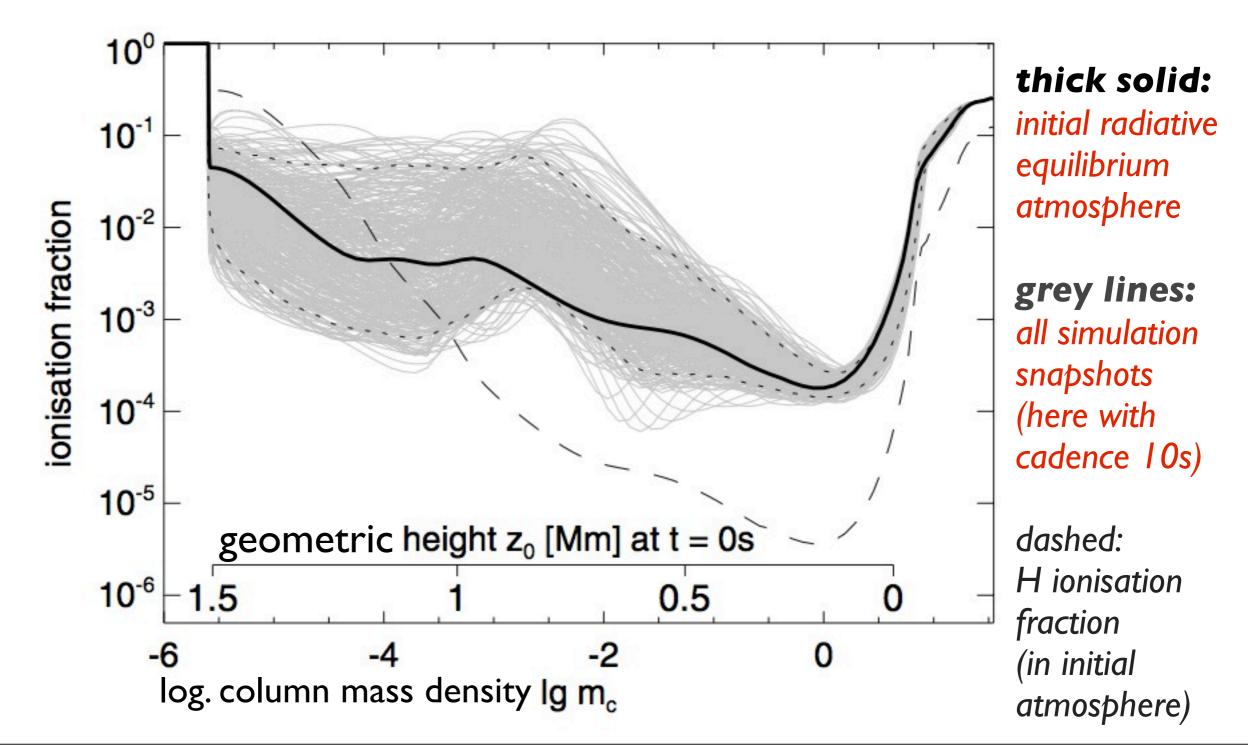




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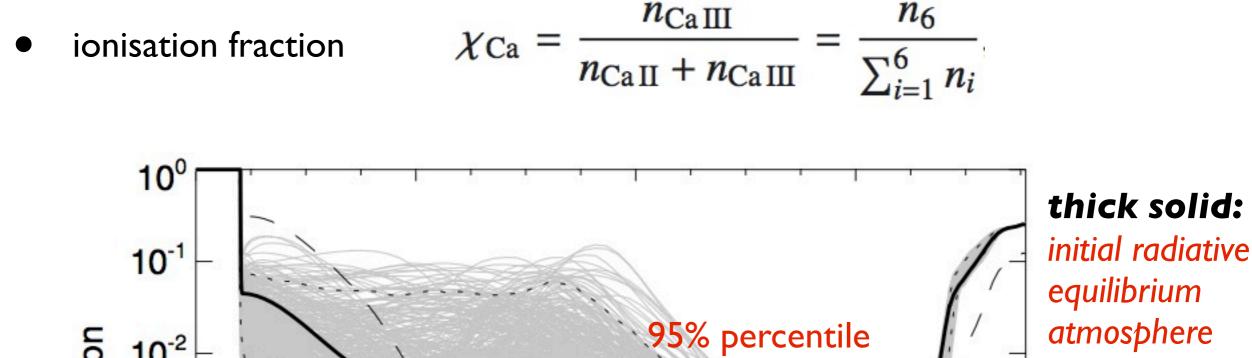
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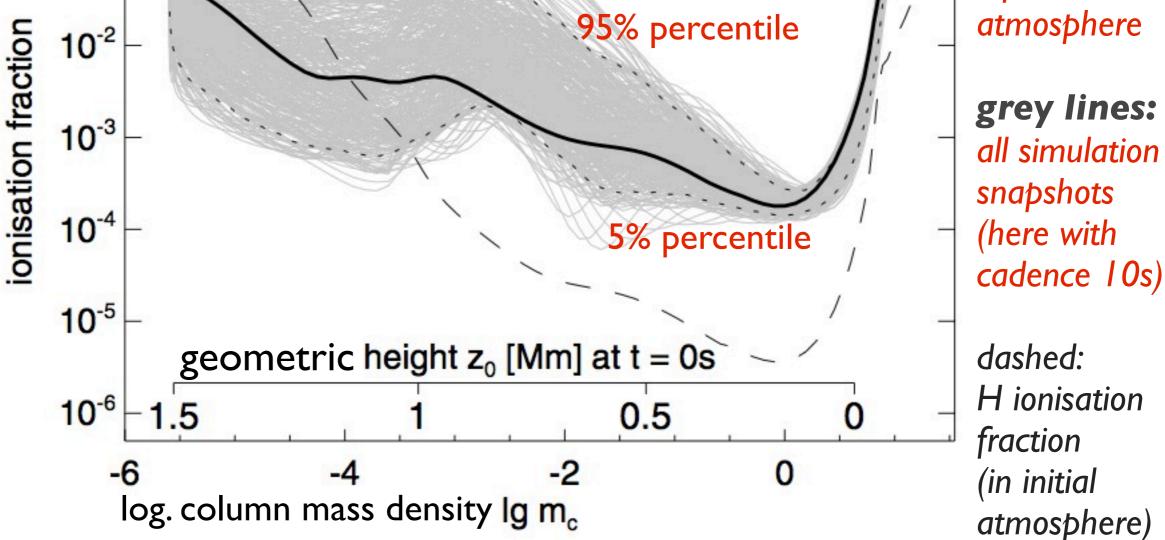
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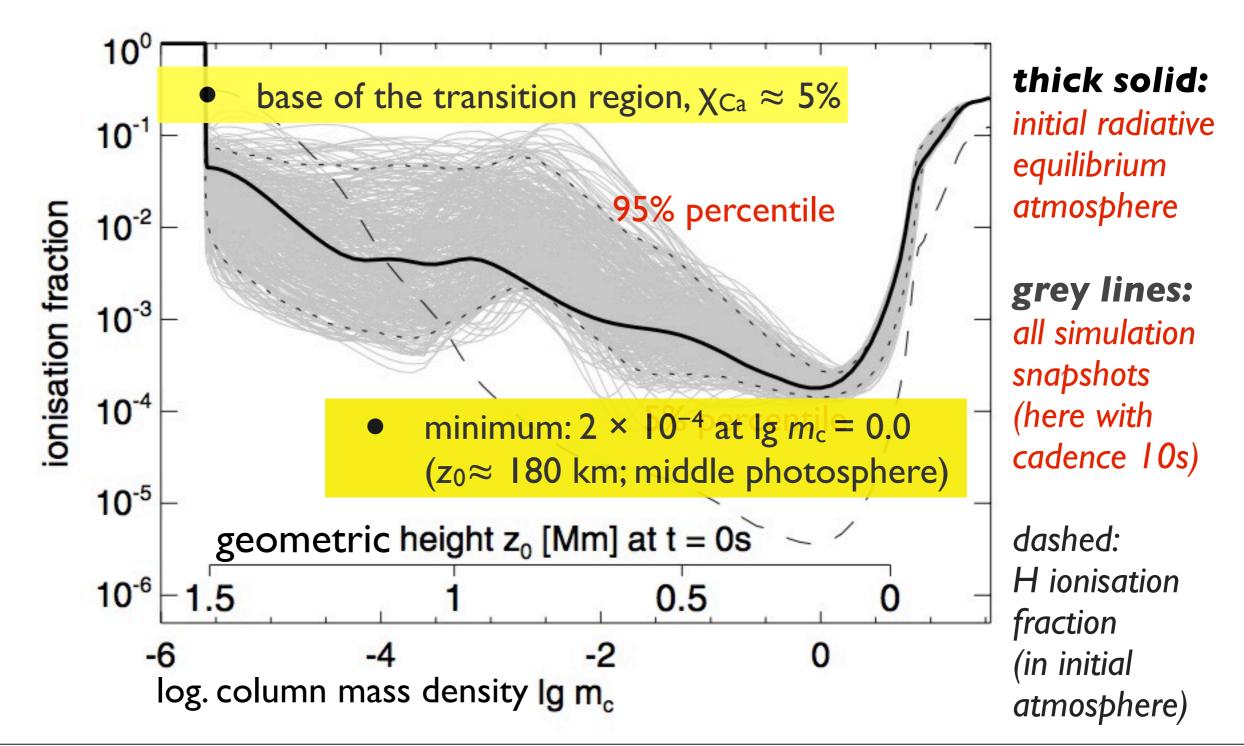




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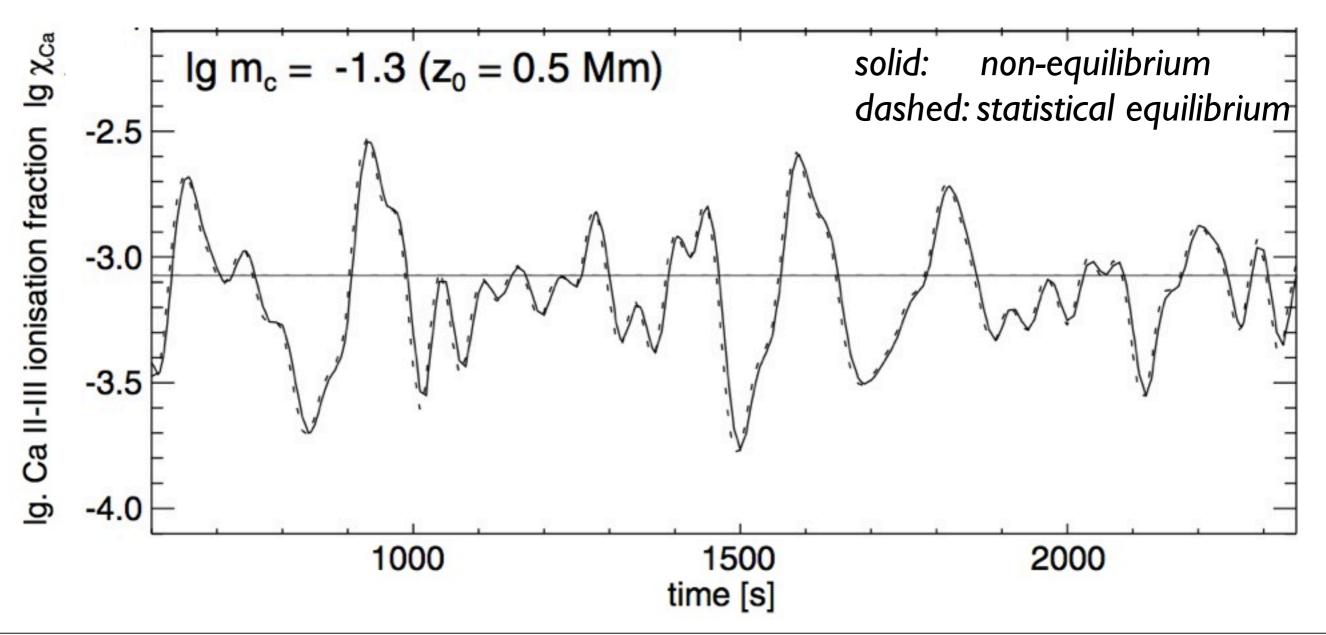
IONISATION FRACTION

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IONISATION FRACTION Temporal evolution at a fixed height top of photosphere/low chromosphere

- Variation of one order of magnitude (min. to max.)
- Essentially <u>no</u> difference between equilibrium and detailed non-equilibrium solution

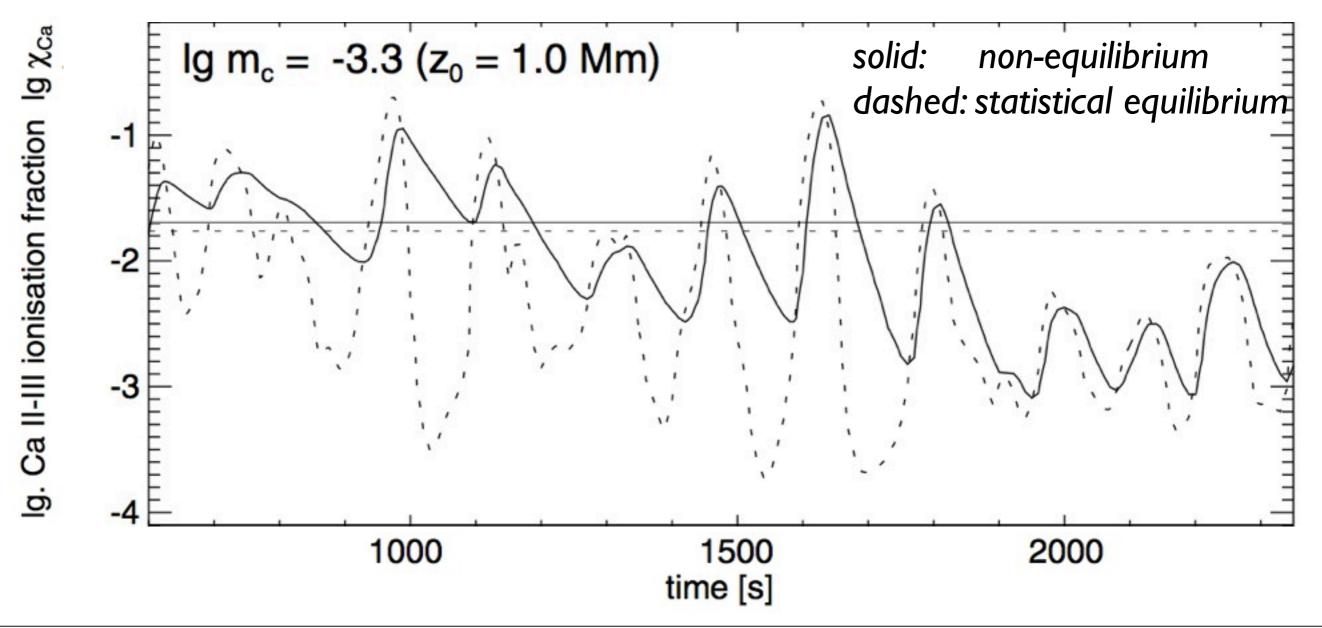


NON-EQUILIBRIUM CALCIUM IONISATION

IONISATION FRACTION Temporal evolution at a fixed height

middle chromosphere

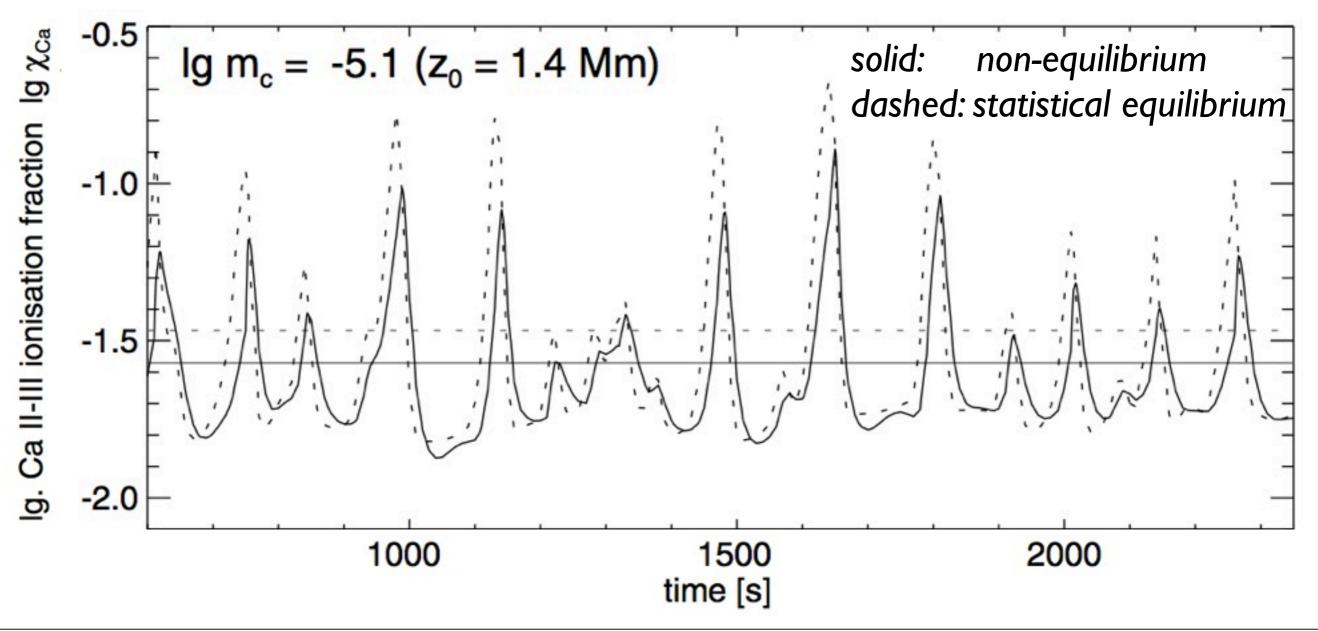
- Variations: two (or more) orders of magnitude
- Non-equilibrium ion. fraction decreases slower after shock wave passage
- Does not reach as small values as in equilibrium case



IONISATION FRACTION Temporal evolution at a fixed height

middle-upper chromosphere

- Variations: up to one order of magnitude
- small differences between equilibrium and non-equilibrium solutions
- non-equilibrium: peaks smaller and with a small time delay (few s)



IONISATION FRACTION Summary

low atmosphere (below the middle photosphere)

- small variations
- minimum of 2×10^{-4} in the middle photosphere
- Statistical equilibrium assumption valid.

chromosphere

- varies strongly in chromosphere due to the passage of shock waves (two orders of magnitude)
- rises with height but less steep than for H
- Relaxes slower after shock wave passage (compared to statistical equilibrium)

transition region

• reaches maximum $\chi_{Ca} \approx 5\%$ at the base of the transition region

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 Relaxation time scale?

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 Relaxation time scale?

Under which conditions is a detailed non-equilibrium treatment necessary?

RELAXATION TIME SCALE

Calculation of the relaxation time scale

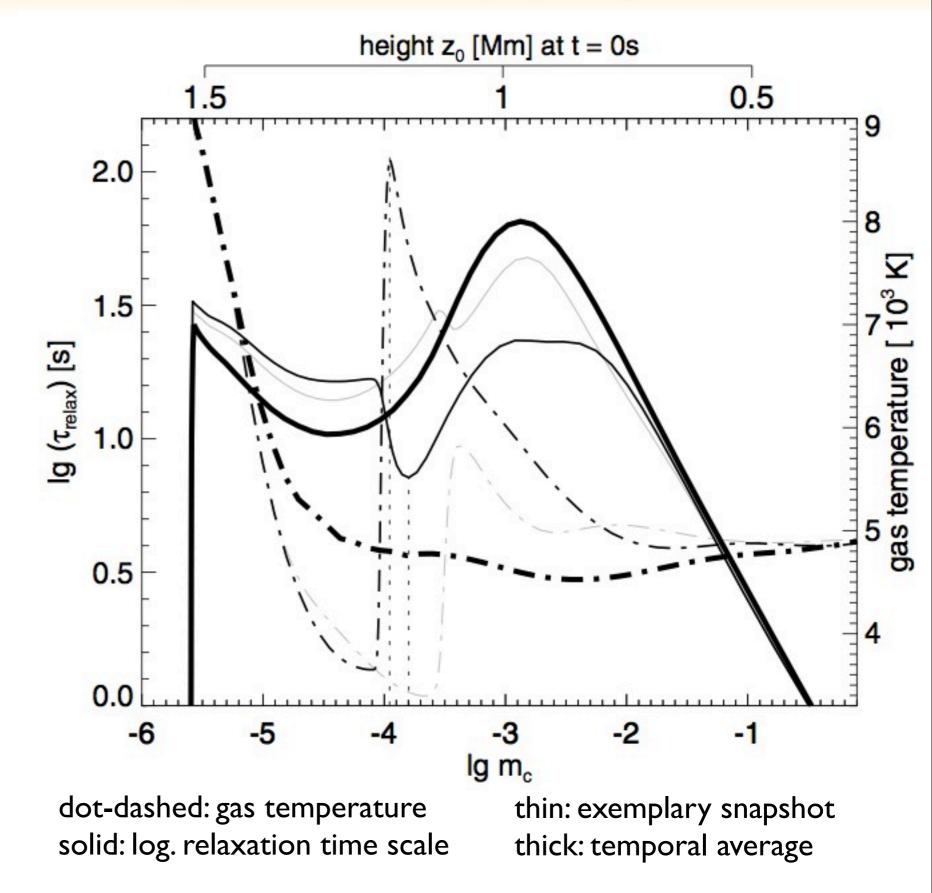
- Method I: numerically from the temporal evolution of an initially perturbed atmosphere
- Method 2: eigenvalue analysis of the rate matrices

RELAXATION TIME SCALE

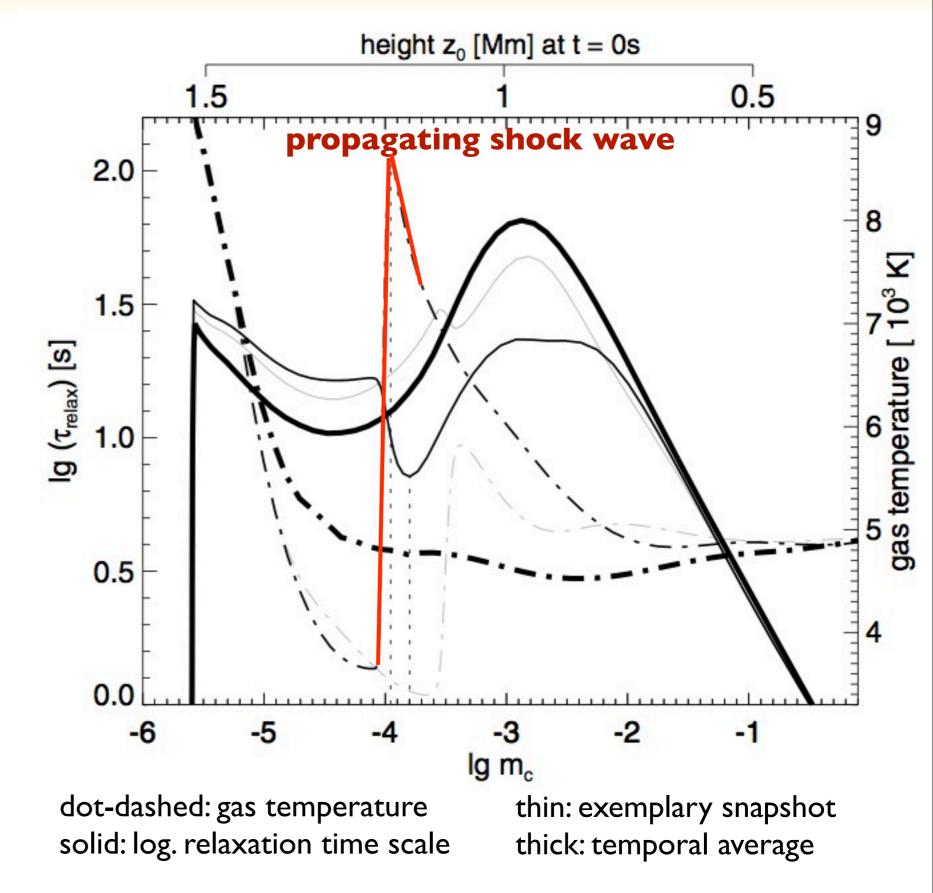
Calculation of the relaxation time scale

- Method I: numerically from the temporal evolution of an initially perturbed atmosphere
- Large number of additional short simulation runs
- Each run starts from a snapshot of the non-equilibrium (TD) sim.
- Initial atmosphere with statistical equilibrium (SE) solution
- Next time step: atmosphere perturbed by increasing the gas temperature by 1%
- Time-evolution followed for 50 s
- In addition, calculation of the SE state of the perturbed atmosphere.

NON-EQUILIBRIUM CALCIUM IONISATION

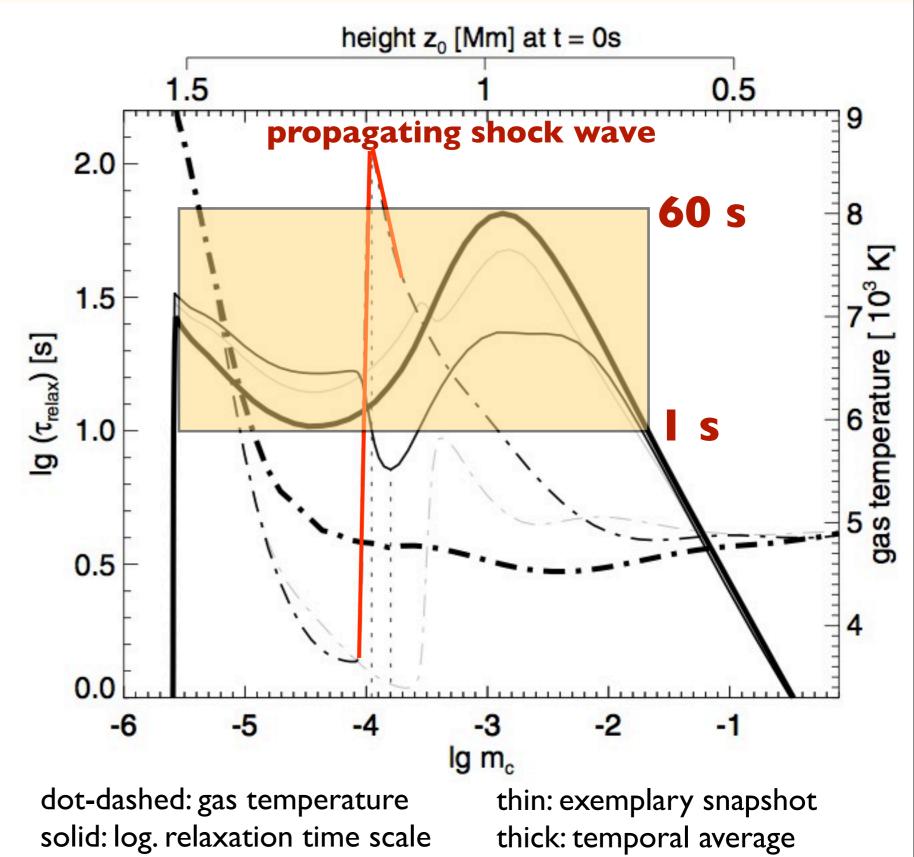


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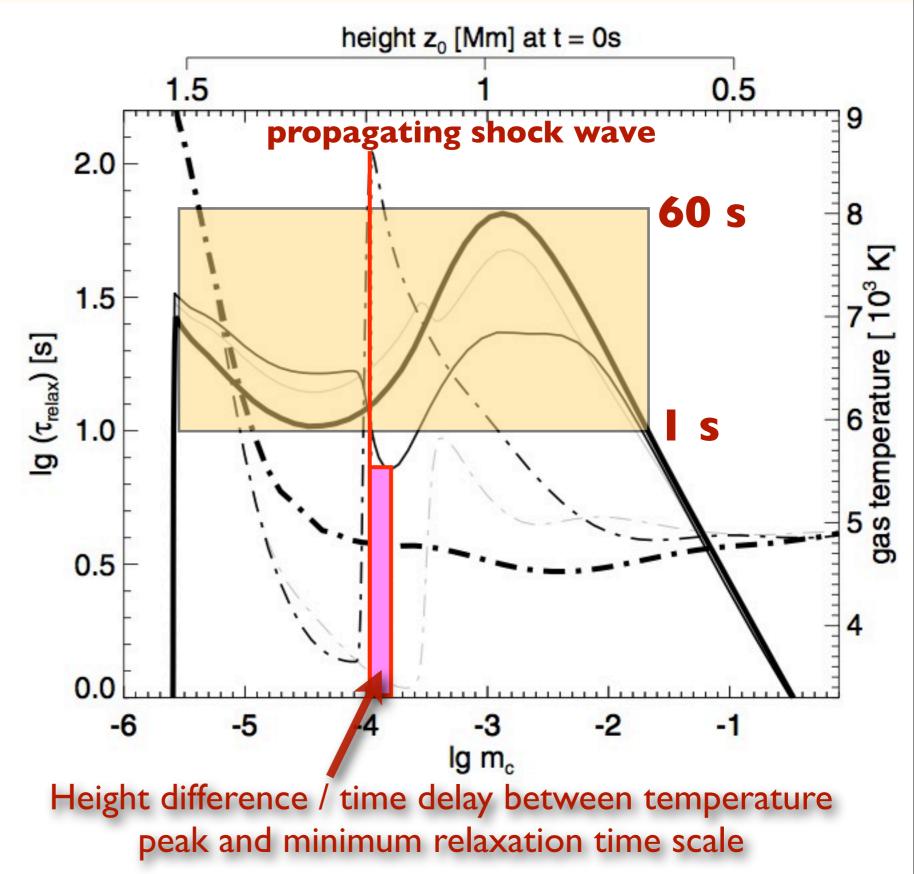
NON-EQUILIBRIUM CALCIUM IONISATION

- <u>photosphere</u>: well below I s
- <u>chromosphere</u>: mostly between Is and 60s
- interval between consequent shock waves ~120 -180s
- ➡ 60s "critical"
- explains the noticeable influence of the nonequilibrium treatment

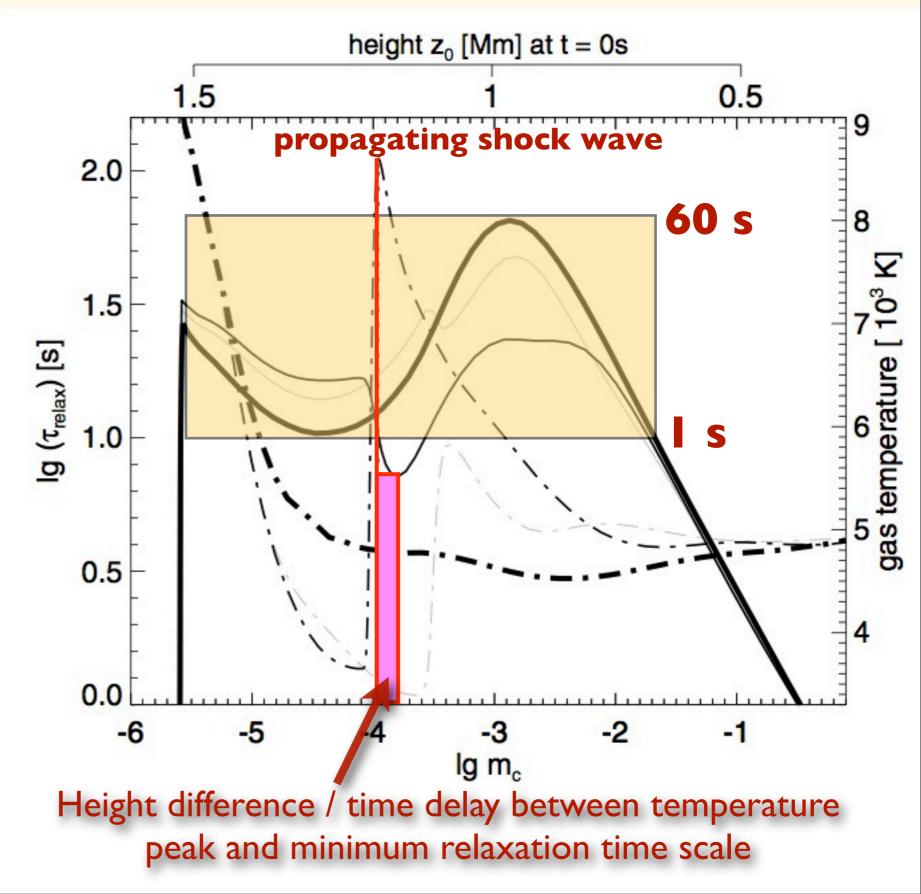


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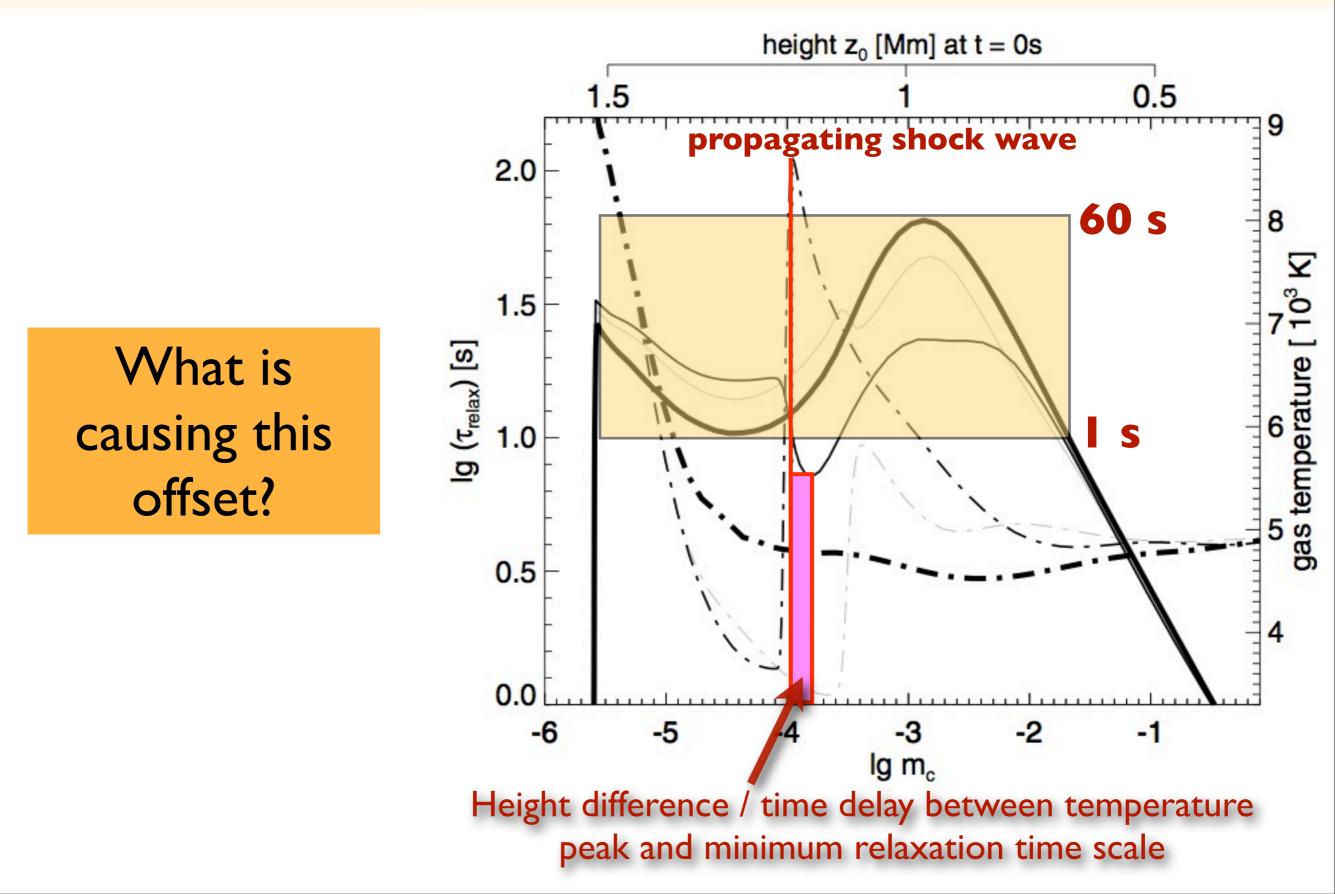
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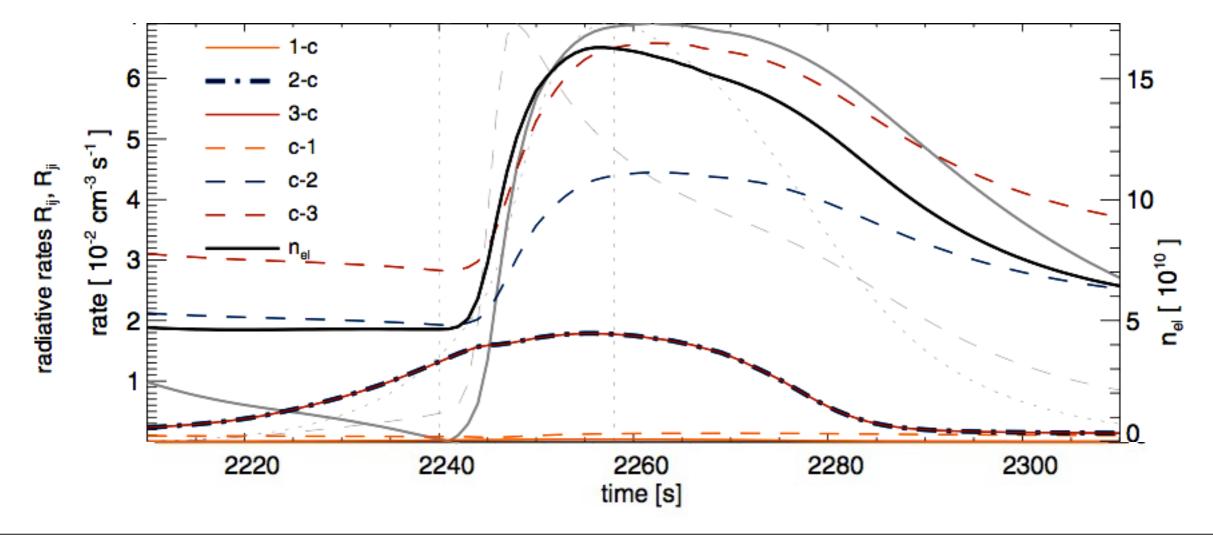
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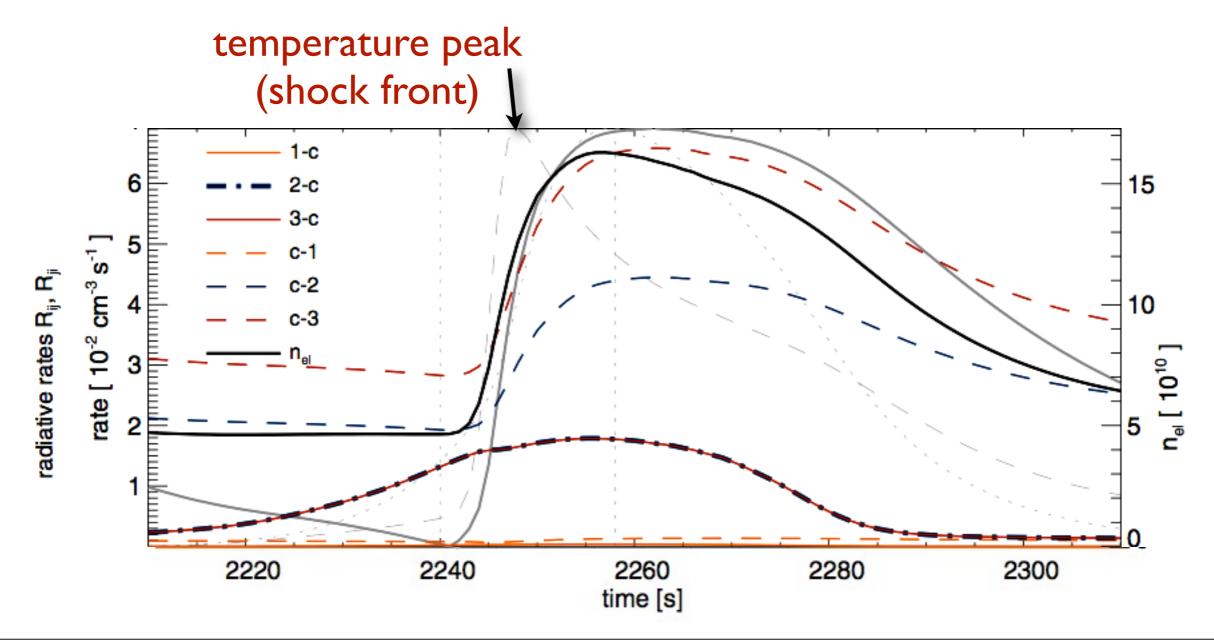
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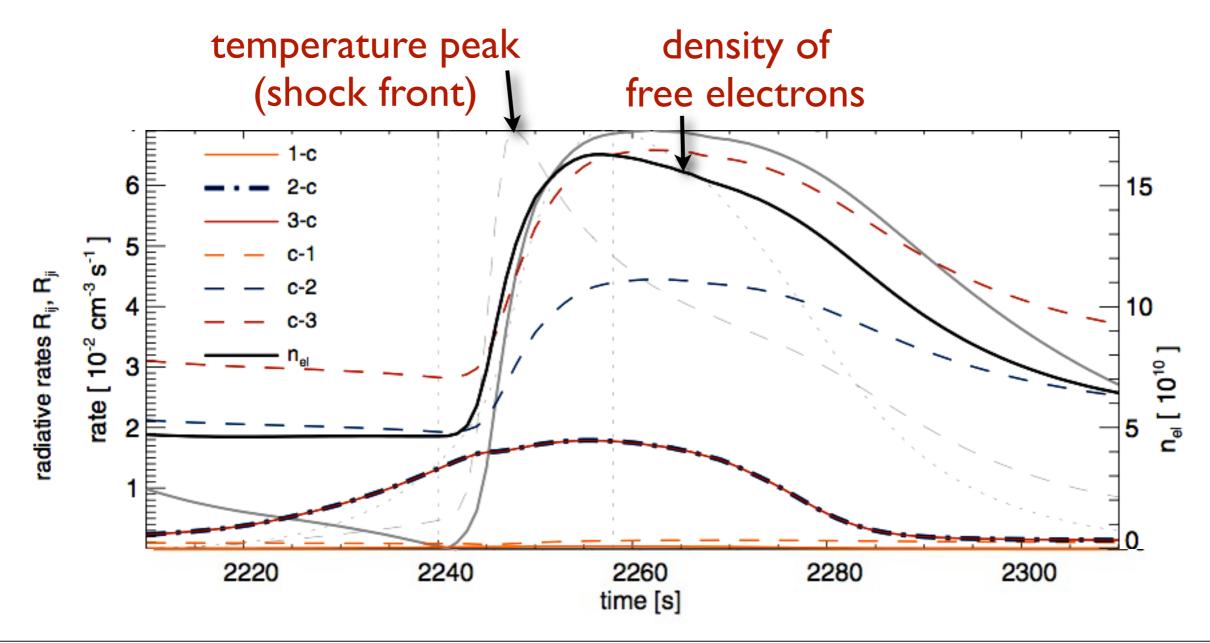
Cause of the time delay?



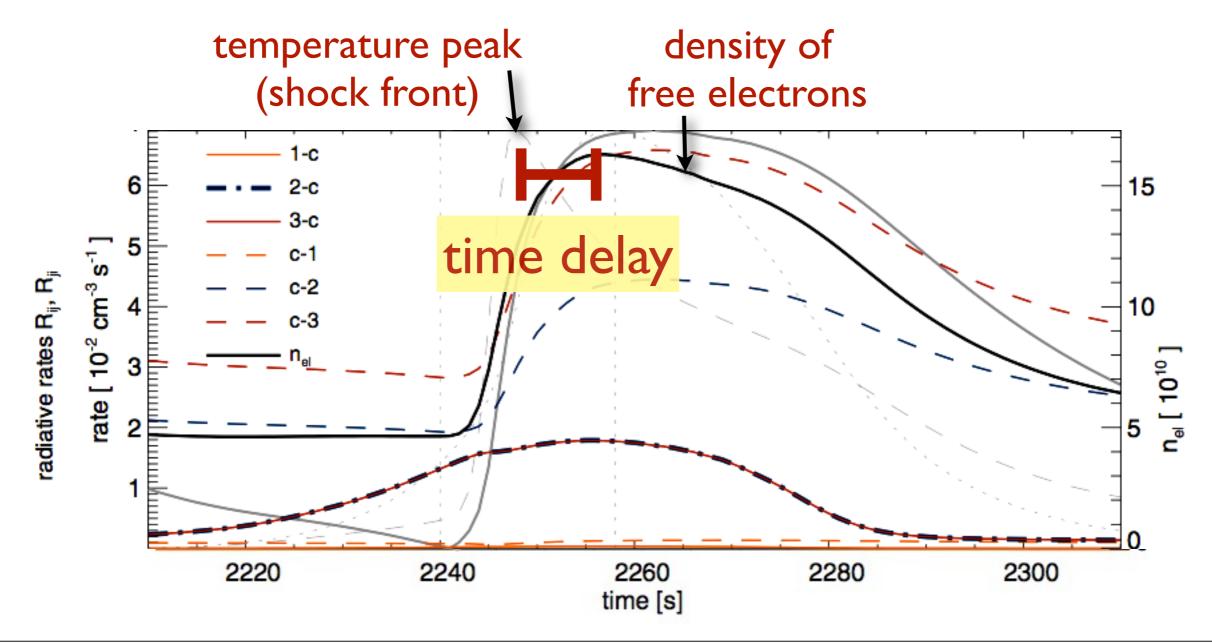
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Cause of the time delay:

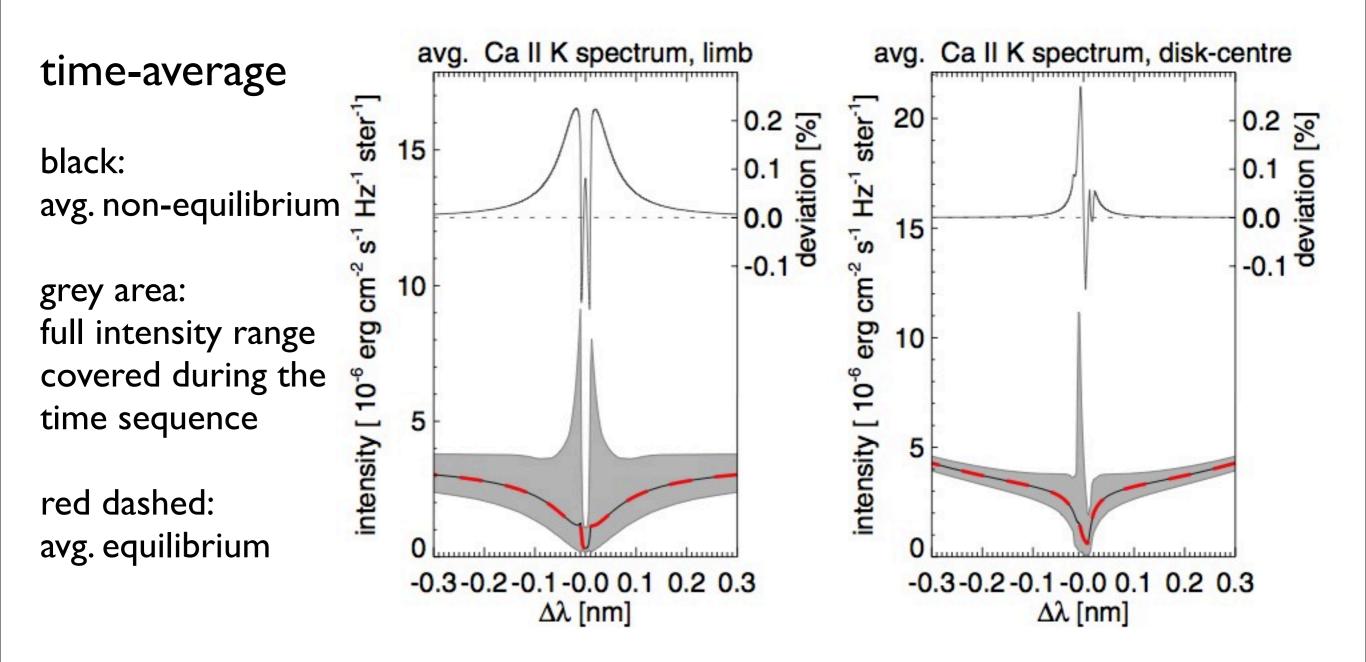
• recombination process involves a collision between a free electron and a Ca III ion (see, e.g., *Mihalas 1978*, p. 130f)

free electrons:

- Ca only a minor electron donor
- Small changes in H ionisation fraction can lead to significant fluctuations of the electron density.
- *H ionisation*/recombination timescale in the chromosphere:
 - on average on the order of one to several hours
 - strongly reduced in hot shock fronts, down to 10 s to 20 s.
 - delayed release of electrons
 - direct effect on the recombination rates of Ca
 - related timescales reach minimum only shortly after the occurrence of temperature peak

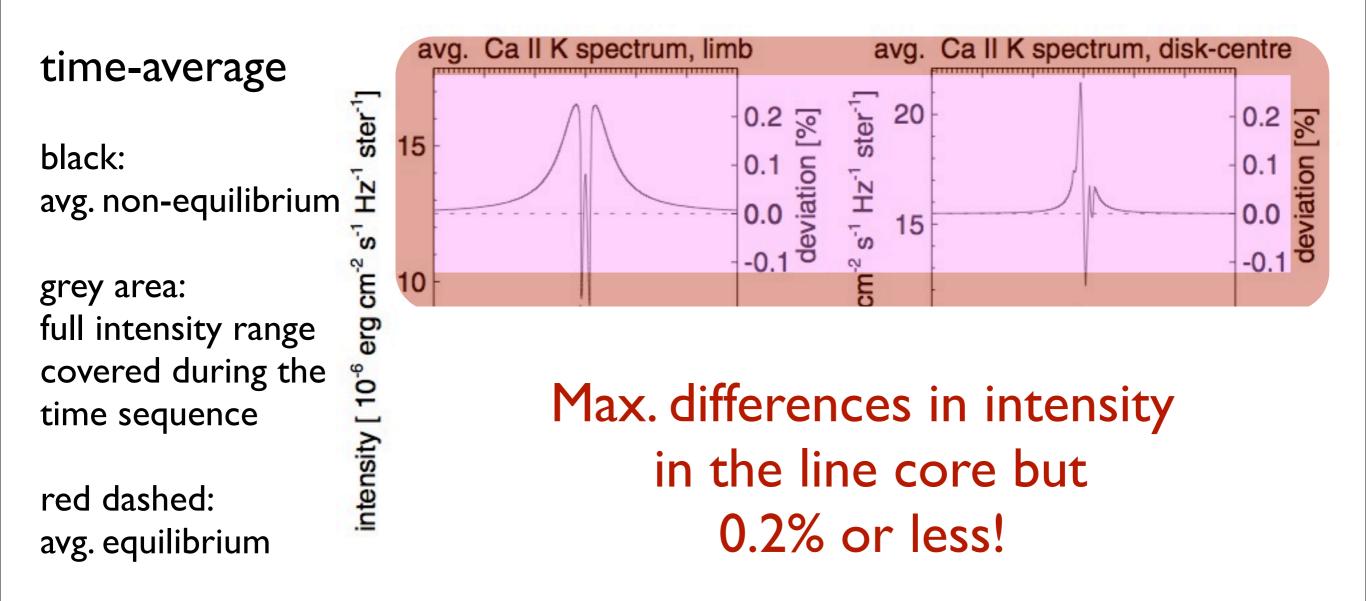
SPECTRUM SYNTHESIS

- Are the non-equilibrium effects measurable?
- Comparison of the spectral line profiles (here Ca K) based on the non-equilibrium and equilibrium solution



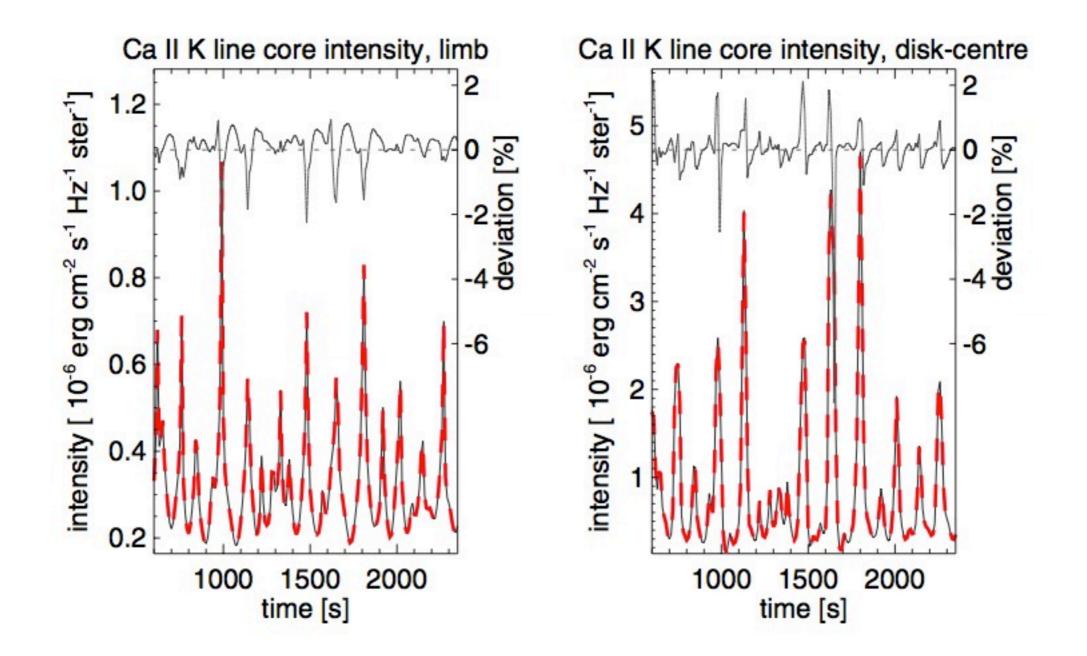
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SPECTRUM SYNTHESIS

- same for the time evolution of the line core intensity:
- differences very small (peaks with a few percent)



CONCLUSIONS

- Ionisation/recombination timescales too long for assumption of instantaneous ionisation equilibrium but not long for the assumption of a constant ionisation fraction (for the considered conditions in the quiet solar chromosphere)
- Ca II-III ionisation fraction generally small
- BUT noticeable deviations from ionisation equilibrium in the middle chromosphere
- error due to assumption of statistical equilibrium is therefore negligible for most applications
- e.g. , effect barely visible in synthesized intensity for the diagnostically important spectral lines of Ca II, i.e., the H and K lines and the infrared triplet